



JEE Main - 6 | JEE 2024

Date: 27/1/2023 Maximum Marks: 300

Timing: 04:00 PM to 07:00 PM

General Instructions

- 1. The test is of **3 hours** duration and the maximum marks is **300**.
- 2. The question paper consists of **3 Parts** (Part I: **Physics**, Part II: **Chemistry**, Part III: **Mathematics**). Each Part has **two** sections (Section 1 & Section 2).
- 3. Section 1 contains 20 Multiple Choice Questions. Each question has 4 choices (A), (B), (C) and (D), out of which ONLY ONE CHOICE is correct.
- **4. Section 2** contains **5 Numerical Value Type Questions**. The answer to each question is an **integer** ranging from 0 to 999.
- 5. No candidate is allowed to carry any textual material, printed or written, bits of papers, pager, mobile phone, any electronic device, etc. inside the examination room/hall.
- **6.** Rough work is to be done on the space provided for this purpose in the Test Booklet only.
- 7. On completion of the test, the candidate must hand over the Answer Sheet to the **Invigilator** on duty in the Room/Hall. **However, the candidates are allowed to take away this Test Booklet with them**.
- 8. Do not fold or make any stray mark on the Answer Sheet (OMR).

Marking Scheme

- 1. **Section 1:** +4 for correct answer, –1 (negative marking) for incorrect answer, 0 for all other cases.
- 2. Section 2: +4 for correct answer, –1 (negative marking) for incorrect answer, 0 for all other cases.

Name of the Candidate (In CAPITALS):	
Roll Number :	
OMR Bar Code Number :	
Candidate's Signature:	Invigilator's Signature

Syllabus

Physics: Rotational Motion, Gravitation, Liquids

Chemistry: Thermochemistry, Thermodynamics, Chemical Equilibrium

Mathematics: Straight Line, Circles, Parabola

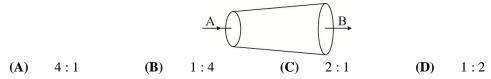
PART - I: PHYSICS

100 MARKS

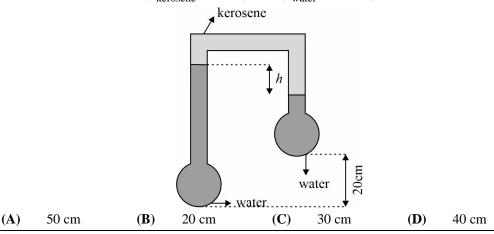
SECTION-1

This section contains 20 Multiple Choice Questions. Each question has 4 choices (A), (B), (C) and (D), out of which ONLY ONE CHOICE is correct.

1. An ideal fluid flows through a pipe of circular cross section with diameters 5 cm and 10 cm as shown. The ratio of velocities of fluid at *A* and *B* is:



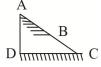
2. For the arrangement shown in the figure the value of h if the pressure difference between the vessel A and B is 3000 N/m^2 is: $(\rho_{\text{kerosene}} = 800 kg / m^3, \rho_{\text{water}} = 1000 kg / m^3)$



- 3. The mass of a hypothetical planet is 1/1000 that of Earth and its radius is 1/8 that of Earth. If a person weighs 500 N on Earth, what would he weigh on this planet?
 - **(A)** 24 N
- **(B)** 48 N
- **(C)** 32 N
- **(D)**

192 N

4. Portion AB of the wedge shown in figure is rough and BC is smooth. A solid cylinder rolls without slipping from A to B. If AB = BC, then ratio of translational kinetic energy to rotational kinetic energy at point C is:



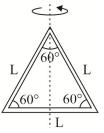
(A) 3/5

(C) 7/5 **(D)** 8/3

5

5. Three identical thin rods, each of mass m and length L are joined to form an equilateral triangle. Find the moment of inertia of the triangle about an axis in its plane perpendicular on one of its sides as shown in the figure.





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- 6. A massless rope is wrapped around a ring (with a groove along its circumference) having a radius R and mass m. The ring is allowed to move downward without slipping on the rope. The linear acceleration of the ring is:

- (A) 2g/3
- **(B)**
- **(C)** *g*/3



7.	When a solid sphere rolls without slipping down an inclined plane making an angle θ with the
	horizontal, the acceleration of its centre of mass is a. If the same sphere slides without friction its
	acceleration is:

- (A) (7/2) a**(B)**
- (5/7) a
- **(C)** (7/5) a
- **(D)** (5/2) a
- 8. The distance between the centres of the Moon and the earth is D. The mass of the earth is 100 times the mass of the Moon. At what distance from the centre of the earth, the gravitational force on a particle will be zero?
- **(B)** $\frac{10D}{11}$ **(C)** $\frac{11D}{10}$ **(D)** $\frac{9D}{10}$
- 9. Water from a tap emerges vertically downwards with an initial velocity V_0 . Assuming pressure is constant throughout the stream of water and the flow is steady, find the distance from the tap at which cross-sectional area of stream is one-fourth of the cross sectional area at the tap.
 - $V_0^2 / 2g$ (A)
- **(B)**
- $15V_0^2 / 2g$ (C) $2V_0^2 / g$
- **(D)** $5V_0^2 / 2g$
- 10. The angle of contact between glass and water is 0° and water (surface tension 70 dyne/cm) rises in a glass capillary up to 6 cm. Another liquid of surface tension 210 dyne/cm, angle of contact 30° and relative density 2 will rise in the same capillary up to:
 - (A) 3 *cm*
- $9\sqrt{3}/2 \text{ cm}$ **(B)**
- **(D)** 6 cm

11.	A planet is moving in an elliptical orbit around the sun. The ratio of minimum distance between planet
	and sun to maximum distance between planet & sun is $\sqrt{3}:2$, if the speed of planet at the minimum
	distance is v_0 , then its speed at the maximum distance is:

- **(A)** $\frac{v_0}{\sqrt{3}}$ **(B)** $\sqrt{3}v_0$ **(C)** v_0 **(D)** $\frac{v_0\sqrt{3}}{2}$
- 12. Two satellites, A and B, have masses m and 2m respectively. A is in a circular orbit of radius R, and B is in a circular orbit of radius 2R around the earth. The ratio of their kinetic energies, T_A/T_B , is:
 - **(A)** $\frac{1}{2}$ **(B)** 1 **(C)** $\sqrt{\frac{1}{2}}$ **(D)** 2
- 13. A wheel, of radius 1m, is rolling purely on a flat, horizontal surface. It's centre is moving with a constant horizontal acceleration = 3m/s^2 . At a moment when the centre of the wheel has a velocity 3m/s, then the acceleration of a point 1/3m vertically above the centre of the wheel will be:
 - **(A)** 10m/s^2 **(B)** 4m/s^2 **(C)** 6m/s^2 **(D)** 5m/s^2
- 14. A uniform ring of radius R is given a back spin of angular velocity $V_0/2R$ and thrown on a horizontal rough surface with velocity of centre equal to V_0 . The velocity of the centre of the ring when it starts pure rolling will be:
 - **(A)** $\frac{V_0}{2}$ **(B)** $\frac{V_0}{4}$ **(C)** $\frac{3V_0}{4}$ **(D)** 0
- 15. A small sphere of volume V falling in a viscous fluid acquires a terminal velocity v_t . The terminal velocity of a sphere of volume 64V of the same material and falling in the same fluid will be:
 - **(A)** $\frac{v_t}{16}$ **(B)** $\frac{v_t}{4}$ **(C)** $16v_t$ **(D)** $4v_t$

- 16. A solid sphere of mass 1kg, radius 10 cm rolls down an inclined plane of height 7m. The velocity of its centre as it reaches the ground level is: (Take $g = 10 \text{m/s}^2$)
 - (A) 7 m/s
- 10 m/s **(B)**
- **(D)** 20 m/s
- **17.** A particle is attached to the lower end of a uniform rod which is hinged at its other end as shown in the figure. Another identical particle moving horizontally collides inelastically and sticks to it. The minimum speed of moving particle so that the rod with particles performs circular motion in a vertical plane will be: [length of the rod is *l*, consider masses of both particle and rod to be same]



- (A)
- **(B)** $\sqrt{10gl}$ **(C)** $\sqrt{\frac{70gl}{3}}$ **(D)**
- A body is projected with a velocity of $2 \times 11.2 \text{ km s}^{-1}$ from the surface of earth. The velocity of the 18. body when it escaped the gravitational pull of earth is:
 - $\sqrt{3} \times 11.2 \text{ km s}^{-1}$

 11.2 km s^{-1}

 $\sqrt{2} \times 11.2 \text{ km s}^{-1}$ **(C)**

- **(D)** $0.5 \times 11.2 \text{ km s}^{-1}$
- 19. A child is standing with folded hands at the centre of a platform rotating about its central axis. The kinetic energy of the system is K. The child now stretches his arms so that the moment of inertia of the system triples. The kinetic energy of the system now is:
 - (A) 2K
- **(B)** $\frac{K}{2}$
- (C) $\frac{K}{3}$
- 3*K*
- A large tank is filled with water (density = 10^3kg/m^3). A small hole is made at depth 10m below water 20. surface. The range of water coming out of the hole is R, on ground. What extra pressure must be applied on the water surface so the range becomes 2R. (Take 1 atm = 10^5 Pa and g = 10 m/s^2)
 - (A) 9 atm
- 4 atm **(B)**
- **(C)**

5 atm

SECTION-2

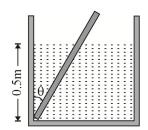
This section contains Five (05) Numerical Value Type Questions. The answer to each question is an integer ranging from 0 to 999 (both inclusive).

1. A string is wound around a hollow cylinder of mass 5 kg and radius 0.5 m. If the string is now pulled with a horizontal force of 40N, and the cylinder is rolling without slipping on a horizontal surface (see figure), then the angular acceleration of the cylinder will be _____ rad/s². (Neglect the mass and thickness of the string)

40N

Two solids P and Q float in water. It is observed that P floats with half of its volume immersed and Q floats with $\frac{2^{rd}}{3}$ of its volume immersed. If the ratio of densities of P and Q is x/4, then x is _____.

- 3. The surface energy of a liquid drop is E. It is sprayed into 729 equal droplets. The work done in spraying is nE. Find n.
- 4. Three particles each of mass m are placed at the corners of an equilateral triangle of side *l*. The work done required to change the side of the triangle from *l* to 3l is $\frac{nGm^2}{2l}$. Find value of *n*.
- 5. A wooden plank of length 1m and uniform cross section is hinged at one end to the bottom of a tank as shown in the figure. The tank is filled with water upto a height of 0.5m. The specific gravity of the plank is 0.5. Find the angle θ (in degrees) that the plank makes with the vertical in equilibrium position. (Excluding the case $\theta = 0^{\circ}$)



PART - II : CHEMISTRY

100 MARKS

SECTION-1

This section contains 20 Multiple Choice Questions. Each question has 4 choices (A), (B), (C) and (D), out of which ONLY ONE CHOICE is correct.

- A mixture of SO_3 , SO_2 and O_2 gases are maintained at equilibrium in 10 litre flask at a temperature at which K_C for the reaction $2SO_2(g) + O_2(g) \Longrightarrow 2SO_3(g)$ is 100. If number of moles of SO_2 and SO_3 are same at equilibrium, the number of moles of O_2 present is:
 - **(A)** 0.01
- **(B)** 0.1
- **(C)** 1
- **(D)** 10
- 2. A sample of Argon gas at 1 atm pressure and 27°C expands reversibly and adiabatically from $1.25\,\text{dm}^3$ to $2.5\,\text{dm}^3$. Calculate the enthalpy change in this process. $C_{V,\,m}$ for Argon is $12.48\,\text{JK}^{-1}\,\text{mol}^{-1}$.

[Given: $(0.5)^{0.66} = 0.633$]

- **(A)** -114.47 J
- **(B)** 110.32 J
- (C) -100.42 J
- **(D)** −126.42 J

3. For the reaction:

 $3Fe_2O_3(s) \rightleftharpoons 2Fe_3O_4(s) + \frac{1}{2}O_2(g), \ \Delta H^0_{298\ K} = 55.5\ kcal \ and \ \Delta G^0_{298\ K} = 46.5\ kcal.$

What is the partial pressure of O_2 at equilibrium at 298 K for this reaction. [Given: $e^{-78} = 1.3 \times 10^{-34}$]

- (A) 1.3×10^{-34}
- **(B)** 2.6×10^{-34}
- (C) 4×10^{-35}
- **(D)** 1.69×10^{-68}

4. For the following equilibrium in a close container.

$$PCl_5(g) \Longrightarrow PCl_3(g) + Cl_2(g)$$

At a constant temperature, the volume of the container is halved suddenly. Which of the following is correct for K_p and α ?

- (A) Both K_p and α changes
- (B) None of the K_p and α change
- (C) K_p changes but α does not
- (**D**) α -changes but K_p does not
- **5.** Which of the following statement is incorrect?
 - (A) Free energy (G) is a path function
 - (B) In the expression G = H TS, TS is the part of the system's energy that is disordered
 - (C) ΔG for a system is equal to $T\Delta S_{Total}$ in magnitude
 - (**D**) ΔG is positive for non-spontaneous reaction
- **6.** Quick lime; (CaO) is produced by heating limestone CaCO₃ to drive off CO₂ gas.

$$CaCO_3(s) \longrightarrow CaO(s) + CO_2(g)$$

$$\Delta H = 180 \,\mathrm{kJ \ mol}^{-1}$$

$$\Delta S = 150 \, \mathrm{JK}^{-1}$$

Assuming that variation of enthalpy change and entropy change with temperature to be negligible, which of the following is correct?

- (A) Decomposition of CaCO₃(s) is always non-spontaneous at room temperature
- (B) Decomposition of CaCO₃(s) becomes spontaneous when temperature is less than 927°C
- (C) Decomposition of $CaCO_3(s)$ becomes non-spontaneous when temperature is greater than $1200^{\circ}C$
- **(D)** All of the above

the temperature to 540 K, AB_3 dissociates into $AB_2(g)$ and $B_2(g)$. If the degree of dissociation of AB_3 is 80%, then final pressure at 540 K is: (A) 5 atm (B) 1.25 atm (C) 10 atm (D) 6.2 atm 8. When 20 g of $CaCO_3$ were put into 10 litre flask and heated to 794°C, 40% of $CaCO_3$ decomposed a equilibrium. K_p for the decomposition of $CaCO_3$ is: (A) 0.7 atm (B) 0.3 atm (C) 1.05 atm (D) 2.5 atm 9. The normal boiling point of a liquid X is 400 K. ΔH_{vap} at normal boiling point is 40 kJ/mol. Select the correct statement. (A) $\Delta S_{vap} < 100 \text{ J/mol K}$ at 400 K and 2 atm (B) $\Delta S_{vap} < 10 \text{ J/mol K}$ at 400 K and 1 atm (C) $\Delta G_{vap} > 0$ at 410 K and 1 atm (D) $\Delta U = 43.22 \text{ kJ/mol K}$ at 400 K and 1 atm (D) $\Delta V_{vap} < V_$	7.	I mole of a gas AB ₃ present in 10 L container at pressure 2.5 atm and 2/3 K temperature. On increasing									
 (A) 5 atm (B) 1.25 atm (C) 10 atm (D) 6.2 atm 8. When 20 g of CaCO₃ were put into 10 litre flask and heated to 794°C, 40% of CaCO₃ decomposed a equilibrium. K_p for the decomposition of CaCO₃ is: (A) 0.7 atm (B) 0.3 atm (C) 1.05 atm (D) 2.5 atm 9. The normal boiling point of a liquid X is 400 K. ΔH_{vap} at normal boiling point is 40 kJ/mol. Select the correct statement. (A) ΔS_{vap} <100 J/mol K at 400 K and 2 atm (B) ΔS_{vap} <10 J/mol K at 400 K and 1 atm (C) ΔG_{vap} > 0 at 410 K and 1 atm (D) ΔU = 43.22 kJ/mol K at 400 K and 1 atm 10. Select the incorrect statement: (A) An adiabatic system can exchange energy with its surroundings (B) A thermodynamic property which is intensive is additive (C) For cyclic process, ΔH is zero 		the ten	nperature to 540	K, AI	B ₃ dissociates	into AB ₂	(g) and	$\mathbf{B}_{2}(\mathbf{g})$. If	the degree o	of dissociation of	
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 (A) 0.7 atm (B) 0.3 atm (C) 1.05 atm (D) 2.5 atm 9. The normal boiling point of a liquid X is 400 K. ΔH_{vap} at normal boiling point is 40 kJ/mol. Select the correct statement. (A) ΔS_{vap} <100 J/mol K at 400 K and 2 atm (B) ΔS_{vap} <10 J/mol K at 400 K and 1 atm (C) ΔG_{vap} > 0 at 410 K and 1 atm (D) ΔU = 43.22 kJ/mol K at 400 K and 1 atm 10. Select the incorrect statement: (A) An adiabatic system can exchange energy with its surroundings (B) A thermodynamic property which is intensive is additive (C) For cyclic process, ΔH is zero 	8.	When	When 20 g of CaCO ₃ were put into 10 litre flask and heated to 794°C, 40% of CaCO ₃ decomposed a								
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 (A) ΔS_{vap} < 100 J / mol K at 400 K and 2 atm (B) ΔS_{vap} < 10 J / mol K at 400 K and 1 atm (C) ΔG_{vap} > 0 at 410 K and 1 atm (D) ΔU = 43.22 kJ / mol K at 400 K and 1 atm 10. Select the incorrect statement: (A) An adiabatic system can exchange energy with its surroundings (B) A thermodynamic property which is intensive is additive (C) For cyclic process, ΔH is zero 	9.	The no	ormal boiling poi	int of a	liquid X is 400) Κ. ΔΗ _{νε}	at nor	mal boiling	point is 40 k	J/mol. Select the	
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 (C) ΔG_{vap} > 0 at 410 K and 1 atm (D) ΔU = 43.22 kJ / mol K at 400 K and 1 atm 10. Select the incorrect statement: (A) An adiabatic system can exchange energy with its surroundings (B) A thermodynamic property which is intensive is additive (C) For cyclic process, ΔH is zero 		(A)	(A) $\Delta S_{\text{vap}} < 100 \text{ J/mol K}$ at 400 K and 2 atm								
 (D) ΔU = 43.22 kJ / mol K at 400 K and 1 atm 10. Select the incorrect statement: (A) An adiabatic system can exchange energy with its surroundings (B) A thermodynamic property which is intensive is additive (C) For cyclic process, ΔH is zero 		(B)	B) $\Delta S_{\text{vap}} < 10 \text{ J/mol K}$ at 400 K and 1 atm								
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 (B) A thermodynamic property which is intensive is additive (C) For cyclic process, ΔH is zero 	10.	Select	the incorrect sta	tement:							
(C) For cyclic process, ΔH is zero		(A)	An adiabatic s	ystem ca	n exchange en	nergy with	its surro	oundings			
		(B)	A thermodynai	mic prop	perty which is	intensive i	s additiv	ve .			
(D) For an isolated system, the entropy can never decrease		(C)	For cyclic proc	ess, ΔF	I is zero						
		(D)	For an isolated	system	, the entropy ca	an never d	ecrease				

11.									
absorbs heat nor releases heat?									
(A) Equilibrium constant will remain constant(B) Equilibrium constant will decrease									
	(D)	Equilibrium constant will decrease Equilibrium constant will increase							
	(D)	Can not be		will illerease					
12.	The co	onversion of	ozone into	oxygen is exot	hermic, un	der what co	ndition ozon	e is more s	stable?
	2O ₃ (g) ==== 3C	$O_2(g)$						
	(A)	At low pre	ssure and lo	ow temperature	e (B)	At high p	ressure and h	igh tempe	rature
	(C)	At high pre	essure and l	ow temperatur	re (D)	At low pr	essure and hi	gh temper	ature
13.	Consi	der the partia	l decompos	ition of A as 2	$2A(g) \rightleftharpoons$	\Rightarrow 2B(g)-	+C(g).		
								300 K. W	hat is the value
	-	for the reac	•			C			
	(A)	40	(P)	$\frac{1}{28}$	(C)	10	(D)	$\frac{28}{10}$	
	(A)	7	(B)	28	(C)	28	(D)	10	
14.	The ed	quilibrium co	nstant K _P	for the reactio	on N_2O_4	g) ==== 2	$NO_2(g)$ is 4	.5.	
	What	would be the	average n	nolar mass (in	g/mol) of	an equilibri	ium mixture	of N_2O_4	and NO ₂ at a
	total p	pressure of 2	atm formed	by the dissoci	ation of pu	ire N_2O_4 ?			
	(A)	69	(B)	57.5	(C)	80.5	(D)	85.5	
				SPACE FO					

Find ΔH at 358K for the reaction $Fe_2O_3(s) + 3H_2(g) \longrightarrow 2Fe(s) + 3H_2O(\ell)$. 15.

Given that, $\Delta H = -33.29\,kJ\,/\,\text{mol}$ at 298 K and C_p for $Fe_2O_3(s),\,Fe(s),\,H_2O(\ell)$ and $H_2(g)$ are 103.8, 25.1, 75.3 and 28.8 J/K mol.

(A) 85.9 kJ/mol **(B)** -28.136 kJ/mol

(C) +28.136 kJ/mol **(D)** 85.9 J/mol

For a reversible adiabatic ideal gas expansion $\frac{dP}{P}$ is equal to: 16.

- **(A)**
- **(B)** $-\gamma \frac{dV}{V}$ **(C)** $\left(\frac{\gamma}{\gamma 1}\right) \frac{dV}{V}$ **(D)** $\frac{dV}{V}$

17. Consider the following reaction:

$$H_2O(\ell) \longrightarrow H_2O(g)$$

$$\Delta H_1 = 44 \,\mathrm{kJ}$$

$$2\text{CH}_3\text{OH}(\ell) + 3\text{O}_2(g) \longrightarrow 4\text{H}_2\text{O}(\ell) + 2\text{CO}_2(g)$$
 $\Delta\text{H}_2 = -1453\,\text{kJ}$

$$\Delta H_2 = -1453 \,\mathrm{k}$$

What is the value of ΔH for second reaction if water vapour instead of liquid water is formed as product?

- $-1409 \, kJ$ **(A)**
- **(B)** $-1629 \, kJ$
- $-1277 \, kJ$ **(C)**
- **(D)** $-1490 \, kJ$

				CDA OF FOR I	20110111	MODIA				
	(A)	1	(B)	2	(C)	3	(D)	4		
	of NO	is added further.	'x' mol	e of NO ₂ is fur	ther pro	duced. What is the	ne value	of equil	librium cons	stant?
	If 1.0	mol of NO is mix	ked with	3.0 mol of NO	3, 'x' n	nol of NO_2 is pr	oduced	at equili	brium. If 2.	0 mol
4 0.	Consider the following reversible reaction $NO + NO_3 \rightleftharpoons 2NO_2$.									
20.	Consid	der the following	reversib	ale reaction NO	+ NO.	2NO.				
	(C)	Twice that of E	3		(D)	Equal to that o	f B			
	(A)	Greater than th	at of B		(B)	Less than that	of B			
	volum	ne V to $V/2$, isoth	ermally	for A and adiaba	tically f	for B. The final p	ressure	of A wil	ll be:	
19. Two samples A and B of an ideal gas, initially at same temperature.							-		-	from
	(A)	40.3 kJ / mol	(B)	43.2 kJ / mol	(C)	4.03 kJ / mol	(D)	3.221	kJ / mol	
	ZII va	•								
	$\Delta H_{\text{vaporisation}}$ of H_2O is: [Density of water = 1g/cc]									
18.	36 mL of pure water takes 100 sec to evaporate from a vessel when a heater of 806 watt is used. The									

SPACE FOR ROUGH WORK

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SECTION-2

This section contains Five (05) Numerical Value Type Questions. The answer to each question is an integer ranging from 0 to 999 (both inclusive).

- Compound X dissociates according to the reaction $2X(g) \rightleftharpoons 2Y(g) + Z(g)$, with degree of dissociation α which is small compared to unity, if the expression for α in terms of equilibrium constant K_P and total pressure P is given as $\alpha = \left(\frac{2K_P}{P}\right)^{1/n}$. The value of n is _____.
- 2. For a equilibrium $A \Longrightarrow B$, $\Delta H = +$ ve, graph between $\log K_{eq}$ and $\frac{1}{T}$ is a straight line with $slope = -\frac{1}{4.606}$. Find ΔH in calories.

3. Calculate the magnitude of resonance energy of N_2O from the following data.

$$\begin{split} [\Delta_f H^0 \ of \ N_2O &= 82.4\,\text{kJ/mol}] \\ BE_{N \,\equiv \, N} &= 946.2\,\text{kJ/mol} \\ BE_{N \,\equiv \, N} &= 418\,\text{kJ/mol} \\ BE_{O \,\equiv \, O} &= 497\,\text{kJ/mol} \\ BE_{N \,\equiv \, O} &= 605.3\,\text{kJ/mol} \end{split}$$

4. At 298 K, if $\Delta_f G^0$ of HCl(g) is 1.72 kJ mol^{-1} , then calculate K_p for the following reversible reaction: $2HCl(g) \Longrightarrow H_2(g) + Cl_2(g)$

[Use: at 298 K :
$$2.303 \text{ RT} = 5700 \text{ J mol}^{-1}$$
 and antilog $(0.6) = 4$]

5. The following equilibrium exists in a closed vessel in 1L capacity.

$$A(g) + 3B(g) \Longrightarrow 4C(g)$$

Initial concentration of A(g) is equal to that of B(g). The equilibrium concentration of A(g) and C(g) are equal. K_c for the reaction is _____.

PART - III: MATHEMATICS

100 MARKS

SECTION-1

This section contains 20 Multiple Choice Questions. Each question has 4 choices (A), (B), (C) and (D), out of which ONLY ONE CHOICE is correct.

- Given the points A(3, 4) and B(4, 3) and the line ax + by + c = 0, $(a, b, c \in R^+)$ and let P_1 and P_2 be lengths of perpendicular from A and B to the line such that $2P_1 + 3P_2 = 10$ then the line ax + by + c = 0 touches the circle.
 - (A) $(x-18)^2 + (y-17)^2 = 4$
- **(B)** $\left(x + \frac{18}{5}\right)^2 + \left(y \frac{17}{5}\right)^2 = 4$
- (C) $\left(x \frac{18}{5}\right)^2 + \left(y \frac{17}{5}\right)^2 = 4$
- **(D)** $(x-3)^2 + (y-6)^2 = 4$
- 2. If the distance between the pair of parallel lines $x^2 + 2xy + y^2 8ax 8ay 9a^2 = 0$ is $25\sqrt{2}$, then a is:
 - (**A**) ±4
- **(B)** ±2
- (**C**) ±3
- **(D)** ± 5
- 3. Consider straight line ax + by = c where $a, b, c \in \mathbb{R}^+$ and a, b, c are distinct. This line meets the coordinate axes at P and Q respectively. If area of $\triangle OPQ = \frac{1}{2}$, 'O' being origin, then:
 - (A) a, b, c are in GP

(B) a, c, b are in GP

(C) a, b, c are in AP

(D) a, c, b are in AP

- Consider the circle $C: x^2 + y^2 = 1$ and the line L: y = m(x+2). If L intersects C at P and Q, then 4. locus of middle point of PQ is:
 - (A) $(x+1)^2 + y^2 = 1$
- **(B)** $x^2 + (y-1)^2 = 1$
- (C) $(x-1)^2 + y^2 = 1$
- $(\mathbf{D}) \qquad x^2 + (y+1)^2 = 1$
- If the equation $ax^2 6xy + y^2 + 2gx + 2fy + c = 0$ represents a pair of lines whose slopes are m and 5. m^2 , then sum of all possible values of a is:
 - (A) -27
- **(B)**
- -19**(C)**
- A light ray gets reflected from the line x = -2. If the reflected ray touches the circle $x^2 + y^2 = 4$ and **6.** point of incidence is (-2, -4), the equation of incident ray is:
 - 4y + 3x + 22 = 0**(A)**

3y + 4x + 20 = 0

4y + 2x + 20 = 0**(C)**

- **(D)** y + x + 6 = 0
- A parabola $y = ax^2 + bx + c$ crosses the x-axis at $(\alpha, 0)$ and $(\beta, 0)$ both to the right of the origin. 7. A circle also passes through these two points. The length of a tangent from the origin to the circle is:
- (B) $\frac{ac}{b^3}$ (C) $\frac{b}{a}$ (D) $\frac{c}{b}$

- Consider a circle, $x^2 + y^2 = 1$ and point $(1, \sqrt{3})$. PAB is a secant drawn from P intersecting circle in 8. A and B (distinct) then range of |PA| + |PB| is:
- **(C)** (0, 4]
- [2, 4]

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9.	If the tangents are drawn from any point on the line $x + y = 3$ to the circle $x^2 + y^2 = 9$, then the chord									
	of co	of contact passes through the point:								
	(A)	(3,5)	(B)	(3,3)	(C)	(5,3)	(D)	(5, 5)		
10.	The e	quation of a st	traight line	joining 2 poi	nts equidist	ant 3 units fro	om a point ((1, 2) lying o	n the circle	
	$x^{2} +$	$y^2 = 5$ is give	en by:							
	(A)	4x + 2y - 1	1 = 0		(B)	2x + 4y -	1 = 0			

- 3x + 2y 1 = 02x + 3y - 1 = 0**(D**) **(C)**
- Line $L_1 = 3x 4y + 1 = 0$ touches the circles C_1 and C_2 whose centers are $A_1(1, 2)$ and $A_2(3, 1)$ 11. respectively, then identify the INCORRECT statement from the following statements.
 - **(A)** L_1 is direct common tangent of these circles
 - $L_{\rm l}$ is transverse common tangent to the circles **(B)**
 - Radius of circle C_1 is $\frac{4}{5}$ units (**C**)
 - Radius of circle C_2 is $\frac{6}{5}$ units **(D)**
- The angle between the chords of the circle $x^2 + y^2 = 100$, which pass through the point (7, 1) and also **12.** divide the circumference of the circle into two arcs whose lengths are in the ratio 2:1, is equal to:
 - **(A)**

- If G is the centroid of triangle with vertices A(a, 0), B(-a, 0) and C(b, c), then $\frac{AB^2 + BC^2 + CA^2}{GA^2 + GB^2 + GC^2} =$ **13.**
 - (A) 1 **(B)**

- 14. The line parallel to the x-axis and passing through the intersection of the lines ax + 2by + 3b = 0 and bx - 2ay - 3a = 0, where $(a, b) \neq (0, 0)$ is:
 - Below the x-axis at a distance of $\frac{3}{2}$ from it (B) Below the x-axis at a distance of $\frac{2}{2}$ from it
 - Above the x-axis at a distance of $\frac{3}{2}$ from it (**D**) Above the x-axis of distance of $\frac{2}{3}$ from it **(C)**
- If the equation $2x^2 + 2hxy + 6y^2 4x + 5y 6 = 0$ represents a pair of straight lines, then the length of **15.** intercept on the x-axis cut by the lines is equal to:
 - (A) 2
- **(B)**
- **(C)**
- **(D)**
- If θ is the angle between tangents to $x^2 + y^2 = 9$ which are normal to the circle **16.** $x^2 + y^2 - 6x - 10y + 33 = 0$ then $\tan \theta =$
- (B) $\frac{5}{3}$ (C) $\frac{8}{15}$ (D) $\frac{15}{8}$
- If $25a^2 + 16b^2 40ab c^2 = 0$, then the line 2ax + by + c = 0 passes through a fixed-point P. Then the **17.** coordinates of *P* can be:
 - - $\left(\frac{5}{2}, 4\right)$ (B) $\left(\frac{5}{2}, -4\right)$ (C) $\left(5, 4\right)$ (D) $\left(5, -4\right)$

- Let P and Q be any two points on the lines represented by 2x-3y=0 and 2x+3y=0 respectively. If 18. the area of triangle *OPQ* (where *O* is origin) is 5, then which of the following is the possible equation of the locus of mid-point of PQ?
 - $4x^2 9v^2 + 15 = 0$ (A)

(B) $4x^2 - 9y^2 - 30 = 0$

 $9x^2 - 4y^2 - 30 = 0$

- **(D)** $4x^2 + 9y^2 30 = 0$
- The range of values of 'a' such that the angle θ between the pair of tangents drawn from (a, 0) to the 19. circle $x^2 + y^2 = 1$ lies in the interval $\left(\frac{\pi}{3}, \pi\right)$ is:

- (A) $(-2, -1) \cup (1, 2)$ (B) $(-\sqrt{2}, -1) \cup (1, \sqrt{2})$ (C) $(-2, -\sqrt{3}) \cup (\sqrt{3}, 2)$ (D) $(-\sqrt{3}, -\sqrt{2}) \cup (\sqrt{2}, \sqrt{3})$
- A focal chord of parabola $y^2 = 4x$ is inclined at an angle of $\frac{\pi}{4}$ with positive x-direction, then the slope 20. of normal drawn at the ends of chord will satisfy the equation.
 - $m^2 2m 1 = 0$ **(A)**

 $m^2 + 2m - 1 = 0$ **(B)**

 $m^2 + m - 1 = 0$ **(C)**

 $m^2 + 2m - 2 = 0$ **(D)**

SECTION-2

This section contains Five (05) Numerical Value Type Questions. The answer to each question is an integer ranging from 0 to 999 (both inclusive).

- 1. If $(\lambda, 2)$ is an interior point of $\triangle ABC$ formed by x + y = 4, 3x 7y = 8 and 4x y = 31 and range of $\lambda = (a, b)$, the value of $\frac{3a + 4b}{4^3 3^2}$ must be _____.
- 2. Find the number of triangles formed by the lines represented by $x^3 x^2 x 2 = 0$ and $xy^2 + 2xy + 4x 2y^2 4y 8 = 0$.
- 3. Slope of a common tangent drawn to 2 curves $x^2 = 4y$ and xy = -2 is equal to ______.
- 4. Consider circles $C_1 \& C_2$ touching both the axes and passing through (4, 4), then the product of radii of these circles is _____.
- 5. If S be the focus of a parabola and PQ be the focal chord, such that SP = 3 and SQ = 6, then the length of latus rectum of the parabola is ______.